

Spectrophotometric method for simultaneous determination of valsartan and substances from the group of statins in binary mixtures

MARIUSZ STOLARCZYK*
ANNA APOLA
ANNA MAŚLANKA
ANNA KWIECIEN
WŁODZIMIERZ OPOKA

*Department of Inorganic
and Analytical Chemistry
Jagiellonian University Medical
College, Faculty of Pharmacy
30-688 Kraków, Poland*

Applicability of derivative spectrophotometry for the determination of valsartan in the presence of a substance from the group of statins was checked. The obtained results indicate that the proposed method may be effective by using appropriate derivatives: for valsartan and fluvastatin – D1, D2 and D3, for valsartan and pravastatin – D1 and D3, for valsartan and atorvastatin – D2 and D3. The method was characterized by high sensitivity and accuracy. Linearity was maintained in the following ranges: 9.28–32.48 $\mu\text{g mL}^{-1}$ for valsartan, 8.16–28.56 $\mu\text{g mL}^{-1}$ for fluvastatin, 14.40–39.90 $\mu\text{g mL}^{-1}$ for atorvastatin and 9.60–48.00 $\mu\text{g mL}^{-1}$ for pravastatin. Determination coefficients were in the range of 0.989–0.999 depending on the analyte and the order of derivative. The precision of the method was high with RSD from 0.1 to 2.5 % and recovery of individual components was within the range of 100 ± 5 %. The developed method was successfully applied to the determination of valsartan combined with fluvastatin, atorvastatin and pravastatin in laboratory prepared mixtures and in pharmaceutical preparations.

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Valsartan (VAL), (S)-3-methyl-2-(N-[[2'-(2H-1,2,3,4-tetrazol-5-yl)biphenyl-4-yl] methyl] pentanamido) butanoic acid, is a cardioprotective drug, specific, selective angiotensin II receptor antagonist (ARB, sartan), type AT₁. Fluvastatin (FLU), (3R,5S,6E)-7-[3-(4-fluorophenyl)-1-(propan-2-yl)-1H-indol-2-yl]-3,5-dihydroxyhept-6-enoic acid, pravastatin (PRV), (3R,5R)-3,5-dihydroxy-7-((1R,2S,6S,8R, 8aR)-6-hydroxy-2-methyl-8-[(2S)-2-methylbutanoyl]oxy)-1,2,6,7,8,8a-hexahydronaphthalen-1-yl)-heptanoic acid, and atorvastatin (ATR), (3R,5R)-7-[2-(4-fluorophenyl)-3-phenyl-4-(phenylcarbamoyl)-5-propan-2-yl-pyrrol-1-yl]-3,5-dihydroxyheptanoic acid, belong to the group of statins (STAT), selective inhibitors of 3-hydroxy-3-methylglutaryl coenzyme A reductase (HMG-CoA reductase). They are commonly used

* Correspondence; e-mail: mariusz.stolarczyk@uj.edu.pl

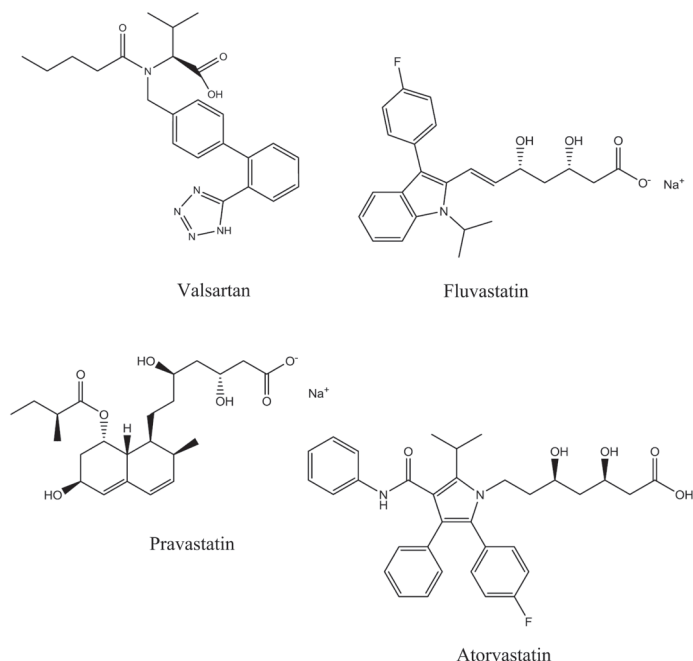


Fig. 1. Chemical structures of valsartan (VAL), fluvastatin (FLU), pravastatin (PRV) and atorvastatin (ATR).

as primary hypolipidemic drugs. Structural formulas of the tested substances are shown in Fig. 1.

Substances from the group of angiotensin II receptor antagonists (ARB) and inhibitors of 3-hydroxy-3-methylglutaryl-coenzyme A reductase are commonly used in the treatment of hypertension with concomitant dyslipidemia. Extensive research on the use of VAL and STAT demonstrated that co-administration of small doses of these substances has a beneficial effect on improving the state of the arteries in patients who have changes associated with age (1), or in the case of changes in the arteries of patients with diagnosed diabetes (2). The beneficial effect of the combination of these substances on the reduction of oxidative stress in combination with anti-inflammatory effect was also observed when compared with the administration of VAL alone (3).

Recent studies of polytherapy with VAL and STAT show that this combination has a synergistic effect and may significantly increase the effectiveness of antihypertensive activity of VAL (4).

In scientific literature, we can find many monographs on the analysis of sartans, determination of individual substances, mixtures or combinations with other substances in laboratory prepared mixtures, pharmaceutical preparations and biological material, usually blood serum or urine. The most commonly used methods for this purpose were chromatographic methods such as HPTLC and TLC (5, 6), LC (7, 8), HPLC (9) and CE (10, 11). The UPLC-MS/MS method was applied for simultaneous determination of telmisartan and hy-

drochlorotiazide in human plasma (12). VAL was also determined by a spectrophotometric method (13, 14). Atorvastatin was determined in tablets by a microwell-based spectrophotometric method (15). Analytical aspects relating to the ARB were also reviewed (16, 17).

Chromatographic methods such as HPLC with spectrophotometric detection were successfully used for the determination of inhibitors of 3-hydroxy-3-methylglutaryl coenzyme A reductase (18–20). Electroanalytical methods (21, 22) and spectrophotometric methods were also applied (23). Pharmacokinetics of statins, their metabolites and impurities were described in review papers by Pasha *et al.* (24) and Ertürk *et al.* (25).

Owing to the undeniable synergistic and beneficial effects of polytherapy with VAL and STAT, pharmaceutical formulations as mixtures of these substances, in the form of two-component medications or polypills, might be expected soon. This was the motif to search for a simple, rapid and accurate method for the determination of these substances in a mixture. We have developed a derivative spectrophotometric method to be successfully applied to the analysis of binary mixtures of VAL and one of the examined statins. In the available literature, we have found no reports on spectrophotometric determination of binary mixtures described in this paper.

EXPERIMENTAL

Instruments

A spectrophotometer UV-VIS Cary 100 (Varian, Australia), Ultrasonic bath Elma (Schmidbauer GmbH, Germany) and centrifuge MPW 54 (MPW Med. Instruments, Poland) were used.

Chemicals

USP reference standards used were: valsartan (Sigma-Aldrich, USA), fluvastatin sodium (USP Convention, Rockville, MD, USA), pravastatin sodium (Sigma-Aldrich), atorvastatin calcium (Sigma-Aldrich).

Reagents and methanol of analytical grade quality were used.

Pharmaceutical formulations

Four pharmaceutical preparations were used: Avasart 160 (manufactured by Polfarmex S.A., Poland, one tablet contains 160 mg of VAL), Lescol 20 (produced by Novartis Pharma GmbH, Germany, one capsule contains 21.06 mg of fluvastatin sodium equivalent to 20 mg of fluvastatin free acid), Pravator 20 (manufactured by Ranbaxy Sp. z o.o., Poland, one tablet contains 20 mg of pravastatin sodium), Atorvasterol 20 (produced by Polpharma SA, Poland, one tablet contains 20 mg of atorvastatin calcium).

Standard and working solutions

Standard solutions of the standard substances of VAL, FLU, PRV and ATR were prepared in methanol.

For direct measurements, standard solutions were diluted with methanol in order to obtain working solutions in the concentration ranges (in $\mu\text{g mL}^{-1}$): 9.28–32.48 for VAL, 8.16–28.56 for FLU, 9.6–48.00 for PRV and 14.40–39.90 for ATR.

Analytical wavelengths for VAL and STAT

Absorption spectra for VAL ($\gamma = 3.92 \mu\text{g mL}^{-1}$), FLU ($\gamma = 20.40 \mu\text{g mL}^{-1}$), PRV ($\gamma = 18.80 \mu\text{g mL}^{-1}$) and ATR ($\gamma = 23.20 \mu\text{g mL}^{-1}$) were recorded in UV using methanol as a blank. Absorption spectrum of VAL shows a poorly marked maximum at 251 nm, FLV and ATR show their maxima close to 251 nm while PRV shows a well-defined maximum at 237 nm. The obtained absorption spectra were converted into first, second and third order derivatives. Using the zero-crossing technique, analytical wavelengths were determined for appropriate derivatives and examined substances.

Method validation

The method was validated according to the ICH recommendations (26) and *Polish Pharmacopoeia* (27).

Linearity. – After transformation of the absorption spectra for VAL, FLU, ATR and PRV in concentration ranges (in $\mu\text{g mL}^{-1}$) of 9.28–32.48, 8.16–28.56, 14.40–39.90 and 9.60–48.00, resp., into derivative curves D1, D2, and D3, the derivative value was read at the specified wavelength and the relationship between the derivative value and concentration was plotted. The following tests were performed in order to verify the model of calibration: Mandel's ($p < 0.05$ indicates the existence of a significant difference between the two models compared – linear and quadratic fit; $p > 0.05$ indicates the equivalence of the two matches), Shapiro-Wilk (value of the statistic test W greater than the critical value allows to assume normal distribution), the Durbin-Watson and the Lagrange tests. The existence of autocorrelation of the residuals was verified.

Accuracy and intra-day precision. – The accuracy of the method was checked by determining the recovery using the standard addition method at three concentration levels (80, 100 and 120 % of analyte concentration). The intra-day precision was checked at 50, 100 and 150 % of analyte concentration. All measurements were repeated three times and the results were calculated using the appropriate regression equation. The obtained results were evaluated statistically.

Limit of detection (LOD) and limit of quantitation (LOQ). – LOD and LOQ were calculated using the following formulae: $LOD = 3.3 S_y/a$ and $LOQ = 10 S_y/a$ (S_y – residual standard deviation of the regression line, a – slope of the calibration line).

Selectivity. – Selectivity of the method with respect to binary mixtures of standard substance solutions was tested by determining the lack of influence of the substance co-present in the mixture on the value of the derivative of the primary component. Selectivity of the method in relation to a pharmaceutical formulation containing one active substance was checked by comparing its absorption spectrum after the active substance extraction process was performed, with the absorption spectrum of the standard solution of similar concentration. No differences confirmed the lack of influence of the excipients.

Laboratory prepared mixtures containing different ratios of VAL/FLU, VAL/ATR and VAL/PRV

Mixtures containing the following ratios of VAL/STAT 2:1; 1:1 and 1:2 were prepared in a series of volumetric flasks by accurately transferring different amounts of working solutions and filling them with methanol to a specified volume. Absorption spectra were then recorded in the range of 200–400 nm. In the next step, absorption spectra were converted into derivatives of an appropriate order and values of derivatives were read at experimentally chosen wavelengths. The appropriate regression equation was used to calculate the concentration of the compound determined.

Application to the analysis of pharmaceutical formulations

VAL and the corresponding STAT were determined in commercially available pharmaceutical preparations (Avasart 160, Lescol 20, Pravator 20, Atorvastrol 20) using the developed method. Ten tablets of an appropriate preparation were powdered in a mortar and a portion of powder corresponding to one tablet was accurately weighed using an analytical balance. Weighed samples were transferred into 10.0-mL flasks and extracted with 10.0 mL of methanol by immersing them in an ultrasonic bath for 30 min. The extract was then centrifuged for 15 min (1500 rpm) and filtered through a syringe filter (0.45 μm). Solutions prepared in this way were then diluted to obtain the concentration of the analyzed substance in the linearity range. VAL and STAT concentrations in the samples were calculated using the appropriate regression equation. Five replicates were made for each pharmaceutical preparation.

RESULTS AND DISCUSSION

Spectral characteristics

None of the tested substances, with the exception of FLV, have a maximum in the range of 270–360 nm that could be used for quantitative analysis. Interference of spectra in the range of 200–280 nm makes direct determination of VAL in the presence of STAT difficult and does not ensure satisfying results according to the ICH requirements for spectrophotometric methods (Fig. 2). Interference of zero-order spectra of analytes significantly impedes direct simultaneous determination of VAL in the presence of FLV, PRV or ATR. Hence, derivative spectrophotometry with the „zero-crossing” technique was used in the course of analysis.

Conversion of zero-order spectra into derivatives causes their significant variation and the appearance of maxima that can be used for quantitative analysis of VAL and STAT mixtures. After evaluation of the first order (D1), second order (D2) and third order (D3) derivatives of the tested spectra and application of the „zero-crossing” method, it was found that in the case of a VAL-FLV mixture the determination may be successfully done using D1, D2 and D3 derivatives (Fig. 3).

In the case of the D2 derivative for the VAL-PRV mixture, it was not possible to determine zeros that could be used for quantitative analysis of the individual components and only D1 and D3 derivatives were used for determination (Fig. 4). Derivatives recorded for

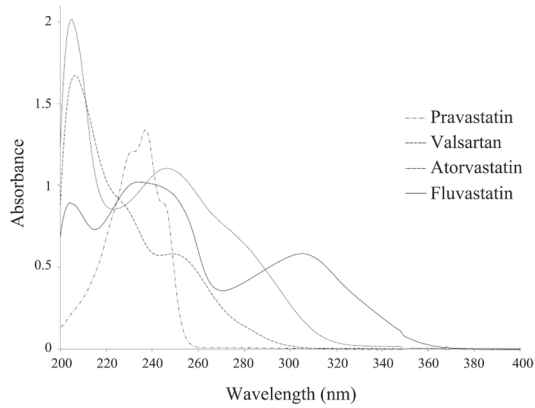


Fig. 2. Zero-order absorption spectra of VAL ($\gamma = 3.92 \mu\text{g mL}^{-1}$), FLU ($\gamma = 20.40 \mu\text{g mL}^{-1}$), PRV ($\gamma = 18.80 \mu\text{g mL}^{-1}$) and ATR ($\gamma = 23.20 \mu\text{g mL}^{-1}$).

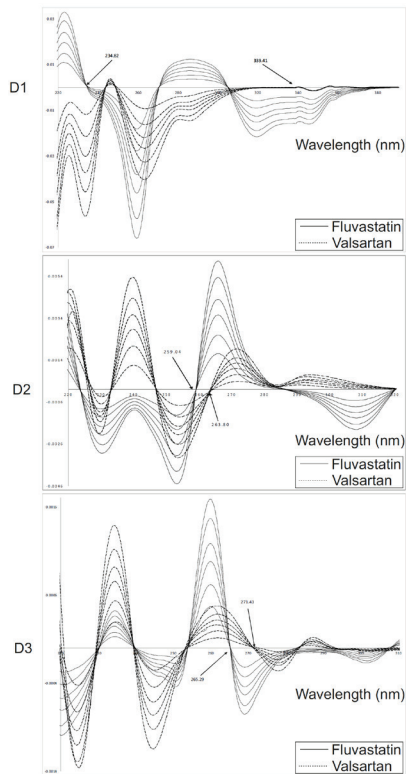


Fig. 3. D1, D2, D3 derivatives for standard solutions of VAL ($\gamma = 9.28, 13.92, 18.54, 23.20, 27.84, 32.48 \mu\text{g mL}^{-1}$) and FLU ($\gamma = 8.16, 12.24, 16.32, 20.40, 24.48, 28.56 \mu\text{g mL}^{-1}$).

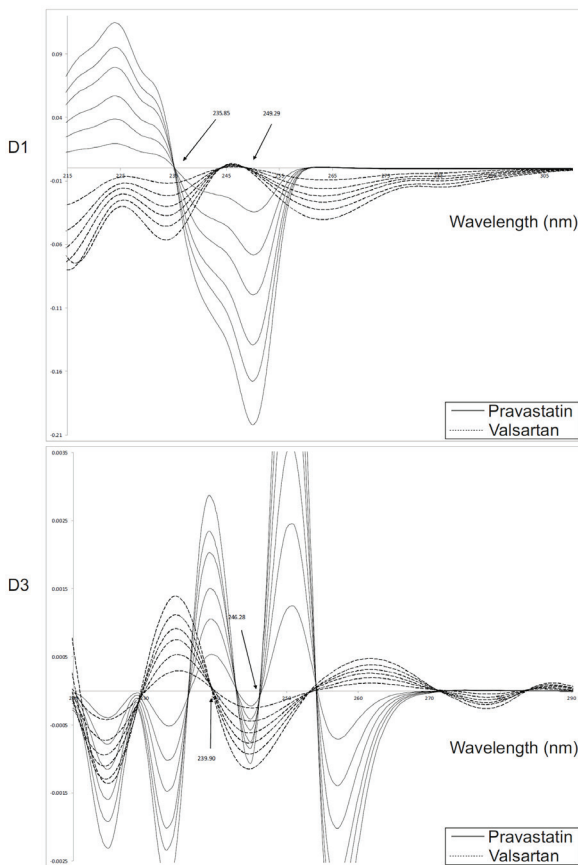


Fig. 4. D1 and D3 derivatives for standard solutions of VAL ($\gamma = 9.28, 13.92, 18.54, 23.20, 27.84, 32.48$ mg μL^{-1}) and PRV ($\gamma = 9.6, 19.2, 28.8, 38.4, 48.0, 57.60$ $\mu\text{g mL}^{-1}$).

the VAL-ATR mixture showed zeros useful for quantitative analysis only in the case of D2 and D3 derivatives (Fig. 5).

Analysis of the obtained derivatives allowed determining appropriate wavelengths. In the case of the determination of VAL and FLV, D1, D2 and D3 derivatives may be used. The following wavelengths were determined using the zero crossing method: D1 $\lambda = 234.82$ nm, D2 $\lambda = 259.04$ nm and D3 $\lambda = 265.29$ nm for VAL and D1 $\lambda = 339.41$ nm, D2 $\lambda = 263.80$ nm and D3 $\lambda = 271.43$ nm for fluvastatin. VAL and PRV can be determined only by using D1 and D3 derivatives, at the wavelength of D1 $\lambda = 235.85$ nm and D3 $\lambda = 246.28$ nm for VAL, and D1 derivative at $\lambda = 249.29$ nm and D3 derivative at $\lambda = 239.90$ nm for PRV. A mixture of VAL and ATR showed zeros in D2 and D3 derivatives and the optimal wavelengths were: D2 $\lambda = 273.20$ nm and D3 $\lambda = 245.96$ nm for VAL, D2 $\lambda = 246.77$ nm and D3 $\lambda = 239.90$ nm for ATR. In addition, the course of the D3 derivatives has demonstrated the possibility to determine ATR or PRV at $\lambda = 239.90$ nm in ternary mixtures, in the presence of VAL and FLV.

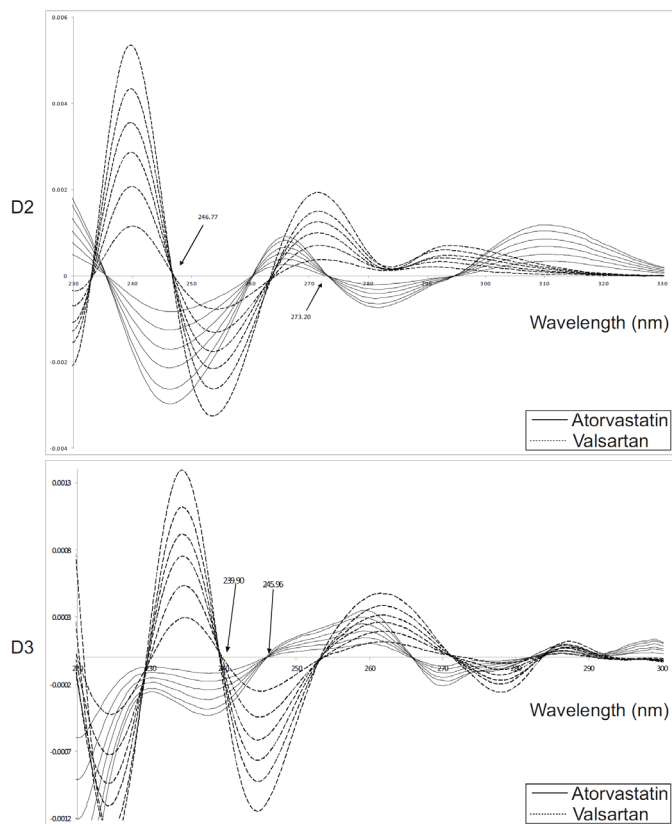


Fig. 5. D2 and D3 derivatives for standard solutions of VAL ($\gamma = 9.28, 13.92, 18.54, 23.20, 27.84, 32.48 \mu\text{g mL}^{-1}$) and ATR ($\gamma = 14.40, 17.10, 22.80, 28.50, 34.20, 39.90 \mu\text{g mL}^{-1}$).

Validation

Validation results together with statistical evaluation are presented in Tables I and II.

Selectivity. – The absorption spectra recorded for solutions of pharmaceutical preparations after extraction did not differ from the absorption spectra of standard solutions. At the selected wavelengths, no additional extremes were observed that could affect the obtained results during analysis.

Linearity. – Linearity was found in the ranges of $9.28\text{--}32.48 \mu\text{g mL}^{-1}$ for VAL, $8.16\text{--}28.56 \mu\text{g mL}^{-1}$ for FLU, $9.60\text{--}48.00 \mu\text{g mL}^{-1}$ for PRV and $14.40\text{--}39.90 \mu\text{g mL}^{-1}$ for ATR. The determination coefficient for examined substances at the chosen analytical wavelengths was $0.9894\text{--}0.9995$. The determined linearity ranges for analytes are given in Tables I and II. The tested concentration ranges provided the optimal absorbance values and allowed obtaining accurate analysis results. In order to verify the calibration model, linear and quadratic fits were

Table I. Validation parameters of the proposed spectrophotometric method for VAL and FLU

Parameter	Valsartan			Fluvastatin sodium		
	D1 ($\lambda = 234.82$ nm)	D2 ($\lambda = 259.04$ nm)	D3 ($\lambda = 265.29$ nm)	D1 ($\lambda = 339.41$ nm)	D2 ($\lambda = 263.80$ nm)	D3 ($\lambda = 271.43$ nm)
Linearity range ($\mu\text{g mL}^{-1}$)		9.28–32.48			8.16–28.56	
Regression equation	$y = 1.80 \times 10^{-3}x - 4.69 \times 10^{-3}$	$y = 6.00 \times 10^{-5}x - 2.20 \times 10^{-5}$	$y = 1.20 \times 10^{-5}x - 2.60 \times 10^{-5}$	$y = 5.52 \times 10^{-4}x - 2.33 \times 10^{-4}$	$y = 1.94 \times 10^{-4}x - 7.20 \times 10^{-5}$	$y = 2.20 \times 10^{-5}x$
Determination coefficient (R^2)	0.9968	0.9955	0.9964	0.9988	0.9990	0.9981
Durbin-Watson test	1.996	2.225	2.017	3.423	3.572	3.367
Shapiro-Wilk test (p)	0.848 (0.152)	0.966 (0.862)	0.876 (0.251)	0.975 (0.924)	0.940 (0.658)	0.928 (0.567)
Mandel's test (p)	0.092 (0.781)	0.341 (0.600)	0.251 (0.651)	0.087 (0.787)	0.081 (0.795)	0.385 (0.579)
LOD ($\mu\text{g mL}^{-1}$)	1.83	2.20	2.75	0.96	0.85	1.50
LOQ ($\mu\text{g mL}^{-1}$)	5.55	6.66	8.33	2.90	2.58	4.54
Accuracy (%) ^a	103.2 ± 1.6	98.8 ± 2.3	102.0 ± 2.0	100.4 ± 1.1	99.6 ± 1.3	100.6 ± 1.2
Precision ^b						
Level 50 %	1.7	2.0	2.3	0.7	2.2	2.2
RSD (%)						
Level 100 %	0.7	0.4	0.1	1.3	1.9	2.5
RSD (%)						
Level 150 %	1.0	0.5	0.3	1.4	1.0	1.1
RSD (%)						

LOD – limit of detection, LOQ – limit of quantitation
^a Accuracy (mean \pm SD) was checked at three concentrations levels: 80, 100 and 120 % of analyte concentration, $n = 3$.
^b Precision was checked at three concentrations levels: 50, 100 and 150 % of analyte concentration, $n = 3$.

Table II. Validation parameters of the proposed spectrophotometric method (VAL, ATR, PRV)

Parameter	Valsartan			Atorvastatin			Valsartan			Pravastatin		
	D2	D3	($\lambda = 273.20$ nm) ($\lambda = 245.96$ nm) ($\lambda = 239.90$ nm) ($\lambda = 246.77$ nm) ($\lambda = 235.85$ nm) ($\lambda = 246.28$ nm) ($\lambda = 249.29$ nm) ($\lambda = 239.90$ nm)	D2	D3	($\lambda = 273.20$ nm) ($\lambda = 245.96$ nm) ($\lambda = 239.90$ nm) ($\lambda = 246.77$ nm) ($\lambda = 235.85$ nm) ($\lambda = 246.28$ nm) ($\lambda = 249.29$ nm) ($\lambda = 239.90$ nm)	D1	D3	D1	D3	D1	D3
Linearity range ($\mu\text{g mL}^{-1}$)	9.28–32.48			14.40–39.90			9.28–32.48			9.60–48.00		
Regression equation	$y = 6.30 \times 10^{-5}x - 2.12 \times 10^{-4}$ $y = 3.50 \times 10^{-5}x - 6.70 \times 10^{-5}$			$y = 8.10 \times 10^{-5}x - 2.14 \times 10^{-4}$ $y = 1.10 \times 10^{-5}x - 3.30 \times 10^{-5}$			$y = 1.70 \times 10^{-3}x - 4.30 \times 10^{-3}$ $y = 3.50 \times 10^{-5}x - 7.20 \times 10^{-5}$			$y = 3.45 \times 10^{-3}x - 2.00 \times 10^{-5}$ $y = 4.90 \times 10^{-5}x + 5.80 \times 10^{-5}$		
Determination coefficient (R^2)	0.9982			0.9977			0.9905			0.9984		
Durbin-Watson test	2.393			2.287			1.898			2.436		
Shapiro-Wilk test (p)	0.796 (0.054)			0.932 (0.593)			0.852 (0.163)			0.891 (0.322)		
Mandel's test (p)	0.659 (0.476)			0.026 (0.881)			3.501 (0.158)			0.029 (0.876)		
LOD ($\mu\text{g mL}^{-1}$)	1.57			1.88			3.67			2.03		
LOQ ($\mu\text{g mL}^{-1}$)	4.76			5.71			11.11			6.14		
Accuracy (%) ^a	102.5 \pm 0.3			101.6 \pm 0.5			102.6 \pm 0.8			101.4 \pm 0.6		
Precision ^b												
Level 50 %	2.0			0.9			0.3			0.3		
RSD (%)												
Level 100 %				1.0			0.3			0.3		
RSD (%)	0.4											
Level 150 %				0.3			0.2			0.1		
RSD (%)	0.4											

LOD – limit of detection, LOQ – limit of quantitation
^a Accuracy (mean \pm SD) was checked at three concentration levels 80, 100 and 120 % of analyte concentration; $n = 3$.
^b Precision was checked at three concentration levels 50, 100 and 150 % of analyte concentration; $n = 3$.

compared by Mandel's test and the results for all cases indicated equivalence of quadratic and linear models. Linear fit was selected for quantitative analyses. The assumption of normal distribution of residuals was confirmed by the Shapiro-Wilk test. The hypothesis of the lack of autocorrelation of random components was verified by the Durbin-Watson test. The results indicated no significant autocorrelation of the residual values for VAL and ATR for all derivatives and PRV D1 ($\lambda = 249.20$ nm). In the case of linearity of FLV (for D1, D2 and D3) and PRV for D3 ($\lambda = 239.90$ nm), the occurrence of significant autocorrelation was observed, but we decided to apply the linear fit because of higher values of R^2 than in the case of quadratic fit. The Lagrange test was used for the third derivative in the case of FLV ($\lambda = 271.41$ nm) and VAL ($\lambda = 246.28$ nm), since the Durbin-Watson test was not conclusive.

Accuracy. – Percent recovery of the analytes at three concentration levels (80, 100 and 120 %) was within 100 ± 5 %. It ranged from 98.8–103.2 for VAL and 99.6–100.6 % for FLU (D1, D2 and D3), 101.4 % (D1) and 101.2 % (D3) for PRV, while it was 102.6 and 101.7 % (D2 and D3, resp.) for ATR.

Sensitivity. – The method was characterized by high sensitivity. LOD was in the range of 1.57 – 3.67 $\mu\text{g mL}^{-1}$ for VAL, 0.85 – 1.50 $\mu\text{g mL}^{-1}$ for FLV, 1.35 – 2.03 $\mu\text{g mL}^{-1}$ for PRV, 3.00 – 3.67 $\mu\text{g mL}^{-1}$ for ATR and the respective LOQs were: 2.86 – 8.33 $\mu\text{g mL}^{-1}$, 2.58 – 4.54 $\mu\text{g mL}^{-1}$, 4.08 – 6.14 $\mu\text{g mL}^{-1}$, 9.09 – 11.1 $\mu\text{g mL}^{-1}$, depending on the class of the derivative.

Analysis of laboratory prepared mixtures

Based on the determined analytical wavelengths for the individual substances and the order of the derivative and using the corresponding regression equations, quantitative analyses of the laboratory prepared mixtures were performed. The composition of mixtures was adjusted so that the VAL:STAT ratio was 2:1; 1:1 and 1:2. In this way, the lack of effect of the second component on the value of the derivative of the examined substance was also proven, even if their relative concentrations were significantly different from each other. This confirms the correct choice of the appropriate analytical wavelength. In the case of VAL-FLU, the obtained recovery was 97.6–102.2 % for VAL and 95.4–103.0 % for FLU, for VAL-PRV 98.8–101.2 % for VAL and 97.6–101.2 % for PRV, for VAL-ATR 98.6–102.4% for VAL and 98.6–102.5 % for ATR. In summary, the obtained results were satisfactorily, ranging between 95 and 102 %. The obtained results together with statistical evaluation are given in Table III.

Analysis of pharmaceutical preparations

The developed and validated method was applied to determination of active substances in pharmaceutical preparations. Results of the measurements of particular active compounds in Avasart 160, Lescol 20, Pravator 20 and Atorvasterol 20 were within the range of 95–105 % of the content declared by the manufacturer. The obtained results were characterized by good precision and accuracy, as evidenced by the standard deviation and 95 % confidence interval presented in Table IV. RSD values were within the range of 0.8–2.1 %. Assay results and statistical analysis are given in Table IV.

Both the validation parameters of the developed method and the results obtained from the assays of the components in mixtures and in pharmaceutical formulations were satisfactory and acceptable according to the ICH requirements for spectroscopic methods

Table III. Determination of analytes in laboratory prepared mixtures by the proposed spectrophotometric method

Concentration ($\mu\text{g mL}^{-1}$)		Recovery (%) ^a		
		Fluvastatin		
		D1 ($\lambda = 234.82 \text{ nm}$)	D2 ($\lambda = 259.04 \text{ nm}$)	D3 ($\lambda = 265.29 \text{ nm}$)
Valsartan	Fluvastatin			
8.12	18.36	100.5 \pm 0.3	102.0 \pm 0.2	102.2 \pm 0.1
20.88	7.14	97.6 \pm 0.8	97.6 \pm 0.9	98.0 \pm 0.4
15.08	13.26	98.0 \pm 0.5	98.8 \pm 0.1	98.3 \pm 0.1
			Valsartan	
		D1 ($\lambda = 235.85 \text{ nm}$)		D3 ($\lambda = 246.28 \text{ nm}$)
Valsartan	Pravastatin			
14.10	38.80	100.2 \pm 0.3	–	101.2 \pm 1.2
18.80	28.80	98.9 \pm 1.0	–	98.8 \pm 1.1
23.50	19.20	101.0 \pm 1.5	–	99.6 \pm 1.5
		Valsartan		Atorvastatin
			D2 ($\lambda = 273.20 \text{ nm}$)	D3 ($\lambda = 245.96 \text{ nm}$)
Valsartan	Atorvastatin			
16.20	33.55	–	99.5 \pm 0.4	98.6 \pm 0.8
29.60	14.80	–	101.7 \pm 1.0	99.4 \pm 0.9
18.65	17.25	–	102.4 \pm 1.1	99.2 \pm 0.3

^a Mean \pm SD, $n = 5$

Table IV. Determination of valsartan, pravastatin, fluvastatin and atorvastatin in pharmaceutical preparations by the proposed method

Pharmaceutical preparation	Content found (mean, $n = 5$) (mg per tablet)			Statistical assessment		
	D1	D2	D3	D1	D2	D3
Avasart 160 (VAL)	159.71	160.36	160.37	SD = 1.3	SD = 1.8	SD = 1.2
				$t \cdot \text{SEM}_{0.95} = \pm 1.6$	$t \cdot \text{SEM}_{0.95} = \pm 2.359$	$t \cdot \text{SEM}_{0.95} = \pm 1.5$
				RSD = 0.8 %	RSD = 1.2 %	RSD = 0.8 %
Pravator 20 (PRV)	20.41	–	20.39	SD = 0.1	–	SD = 0.2
				$t \cdot \text{SEM}_{0.95} = \pm 0.2$		$t \cdot \text{SEM}_{0.95} = \pm 0.2$
				RSD = 2.0 %		RSD = 1.8 %
Lescol 20 (FLV)	21.99	21.03	21.88	SD = 0.8	SD = 0.4	SD = 0.4
				$t \cdot \text{SEM}_{0.95} = \pm 1.0$	$t \cdot \text{SEM}_{0.95} = 0.2$	$t \cdot \text{SEM}_{0.95} = \pm 0.5$
				RSD = 1.5 %	RSD = 1.9 %	RSD = 2.0 %
Atorvasterol 20 (ATR)	–	20.45	20.89	–	SD = 1.0	SD = 0.9
					$t \cdot \text{SEM}_{0.95} = \pm 0.2$	$t \cdot \text{SEM}_{0.95} = \pm 0.3$
					RSD = 1.0 %	RSD = 1.2 %

$t \cdot t \cdot \text{SEM}_{0.95}$ – margin of error

(26). The previously proposed spectrophotometric methods for the determination of valsartan (12) in a narrower linearity range showed similar RSD and *LOD* values. In this case, the analysis was using only the first-order derivative, which in some cases can significantly complicate the analysis process. The possibility of selection of the derivative allows optimizing the obtained results and selecting the appropriate procedure for a mixture of defined composition. Another procedure described in the literature was the spectrophotometric method that requires determination of the isosbestic point but the application of this procedure may be limited when there is no possibility to determine that point (13). In the case of determination of statins in pharmaceutical preparations by a spectrophotometric method, Ergin *et al.* (21) used the reaction of derivatization of statins with 7,7,8,8-tetracyanoquinodimethane, which, on the one hand, allowed the use of the VIS analysis but, on the other hand, significantly complicated the analysis process compared to the method using the zero-crossing technique. The application of separation techniques such as HPLC (22) gives a much larger linearity range and better validation parameters (method applied for the determinations in human serum) but when it was used for the assay in pharmaceutical formulations, slightly higher RSD values were obtained than with the spectrophotometric method described above. An additional advantage of the spectrophotometric method is the low cost and short time of the analysis. In the case of the HPLC method, the separation of examined substances on Inertsil ODS 3V column (4.6 × 250 mm, 5 μm) took about 40 minutes (22) while the time of spectrophotometric analysis that we propose does not exceed 1 min.

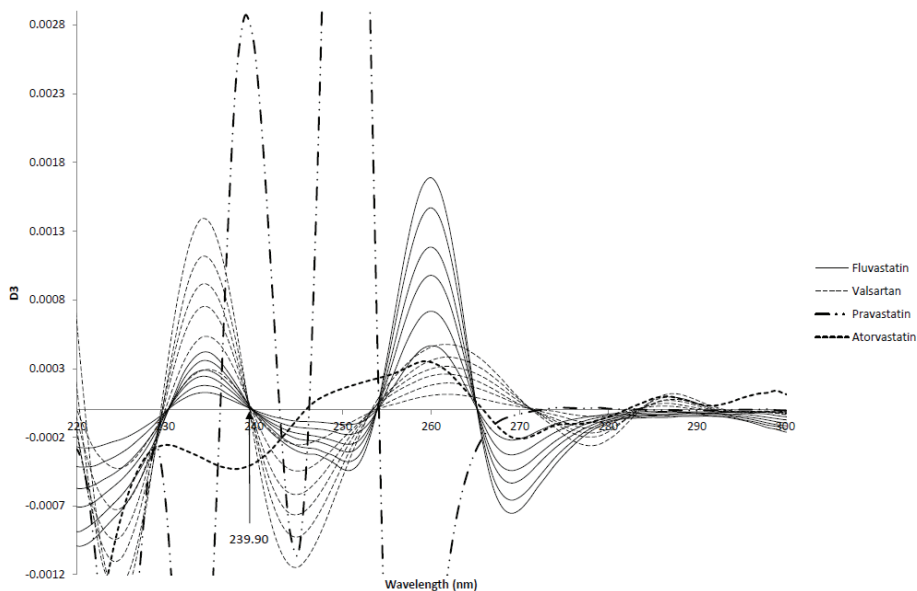


Fig. 6. D3 derivatives for standard solutions of VAL, FLV, ATR and PRV.

Other applications

Analysis of the derivatives showed that the third derivative of the absorption spectra of FLV and VAL showed zero at $\lambda = 239.90$ nm. At this wavelength for PRV and ATR, the extremes occurred that are likely to be used for quantitative analysis of ternary mixtures containing FLV, VAL and ATR or FLV, VAL and PRV (Fig. 6).

CONCLUSIONS

The spectrophotometric method for simultaneous determination of a substance from the group of angiotensin receptor blockers, valsartan, and statins (fluvastatin, pravastatin or atorvastatin) was assessed. Achieving reliable results is possible by derivatization of zero-order spectra, using the „zero-crossing“ method and selecting the appropriate order of the derivative and wavelength. Fulfilment of these conditions means that the method used is accurate, precise, sensitive and specific, as evidenced by the statistical parameters.

Based on the conducted measurements and statistical evaluation of the obtained results, it can be concluded that the proposed procedures can be useful for quantitative analysis of valsartan and statins both in model mixtures and in pharmaceutical preparations, being an alternative to expensive and time-consuming separation techniques.

The proposed method is simple, rapid, selective and sensitive and may be an alternative to chromatographic methods.

Acronyms/abbreviations. – VAL – valsartan, ATR – atorvastatin, FLV – fluvastatin, PRV – pravastatin, HMG-CoA – 3-hydroxy-3-methylglutaryl coenzyme-A reductase, STAT – statins, ARB – angiotensin receptor blocker.

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